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in Schroeder), DH H 2O = 285:83 kJ mol 1 and DH = 890:36 kJ mol 1 (12) This time, the final volume is 1 3 of the initial volume, since the 2 moles of water has condensed out as liquid with negligible volume compared to the gases. Thus the atmosphere does work PDV = RTDn (13) = 8:31 J K 1 (298 K)(2) (14) = 4:953 kJ (15)

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