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Chapter 13 - Gases 195 Exercise 13.3 - Equation Stoichiometry: Iron is combined with carbon in a series of reactions to form pig iron, which is about 4.3% carbon.
 $2C + O_2 \rightarrow 2CO$ $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$ $2CO + C \rightarrow 2C$ (in iron) CO_2 Pig iron is easier to shape than pure iron, and the presence of carbon lowers its melting point

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mix and form solutions. To show how you can predict whether substances are soluble in water.

Chapter 13 Solution Dynamics - An Introduction to Chemistry

Name _____ Date _____ Class 76 Chemistry: Matter and Change • Chapter 13 Study Guide for Content Mastery Section 13.3 Liquids and Solids In your textbook, read about liquids and solids. In the space at the left, write true if the statement is true; if the statement is false, change the italicized word or phrase to make it true.

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- State the relationships among pressure, temperature, and volume of a constant amount of gas.

Chemistry: Matter and Change

196 Study Guide for An Introduction to Chemistry Chapter 13 Map Chapter Checklist Read the Review Skills section. If there is any skill mentioned that you have not yet mastered, review the material on that topic before reading this chapter. Read the chapter quickly before the lecture that describes it.

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smallest unit of an element that still maintains the properties that element.
ELEMENT- a pure substance made up of only one type of atom. MATTER and CHANGE Chapter 1 Section 1 Chemistry (12th Edition) answers to Chapter 2 - Matter and Change - 2.1 Properties of Matter - 2.1 Lesson

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