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Conversion Reactions
Of Organic

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~~Haloalkanes /u0026~~

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?

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and Haloarenes
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Example 10.2 Intext

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Haloalkanes

Haloarenes | exercise

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solution part-1

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Haloalkanes And

Haloarenes

Haloarenes are less polar than

haloalkanes and are insoluble in water.

This is because of lack of hydrogen bonding. As a result, the attractive forces in haloarenes—water

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system remain less than the attractive forces in H₂O

molecules which are hydrogen bonded.

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Chemistry

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Haloalkanes and

Haloarenes Question

10.1: Name the

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following halides
according to IUPAC
system and classify
them as alkyl, allyl,
benzyl (primary,
secondary, tertiary),
vinyl or aryl halides :

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Haloalkanes and
Haloarenes - Study

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Haloalkanes Class 12

Notes Chapter 10 The
bond (carbon-halogen

bond) of alkyl halide

is polarised. Halogen

atom carries a partial

negative charge

Carbon atom carries a

partial positive

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Haloalkanes are classified as fluoro, chloro, bromo or iodo compounds according to the type of halogen present and as mono-, di- tri-, tetra-haloalkanes, etc., according to the one, two, three, four, etc.,

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halogen atoms
respectively present
in their molecule. 2.

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Notes Chemistry
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Chemistry Notes for
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Haloalkanes and
Haloarenes. The
replacement of
hydrogen atom(s) in

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hydrocarbon,
aliphatic or aromatic,
by halogen atom(s)

results in the
formation of alkyl
halide (haloalkane)
and aryl halide
(haloarene),
respectively.

Classification of
Halogen Derivatives.

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Haloalkanes and ...

Haloalkanes are the halogen derivatives of hydrocarbons and

these are of the

following types

depending upon the number of hydrogen

atoms present in them. Haloarenes are

aromatic compounds in which the halogen

atom is attached directly to the carbon

atoms.

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atom of the aromatic ring. The solutions of chapter 10 also give exemplary problems, HOTS (Higher Order Thinking Skills), MCQs exercises, assignments and worksheets which will help you to learn the concept comprehensively and to make haloalkanes

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replacement of

hydrogen atom (s) in

a hydrocarbon,

aliphatic or aromatic,

by halogen atom (s),

results in the

formation of alkyl

halide. (haloalkane)

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and aryl halide

(haloarene),
respectively.

Haloalkanes contain halogen atom (s) attached to the sp^3 hybridized carbon atom of an alkyl group whereas haloarenes contain halogen atom (s) attached to sp^2 hybridized carbon atom (s) of an aryl

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Haloalkanes and

Haloarenes. Nature of

C-X bond in alkyl

halides: X is more

electronegative than

carbon. So, the C-X

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bond is polarized with C having a partial positive charge and X having a partial negative charge.

Preparation of haloalkanes:

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Haloalkanes II: 3:

Properties of

Haloalkanes I: 4:

Properties of

Haloalkanes II: 5:

Properties of

Haloalkanes III: 6:

Allyl Halides and

Dihalides I: 7:

Reactions Of

Dihalides I: 8:

Preparation of

Haloarenes I: 9:

Properties of

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Haloalkanes I: 10:

Solved Problem Set I:

11: Solved Problem

Set II: 12 ...

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Notes 1. Haloalkanes
are classified as

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fluoro, chloro, bromo or iodo compounds according to the type of halogen present and as mono-, di- tri-, tetra- haloalkanes, etc., according to the one, two, three, four, etc., halogen atoms respectively present in their molecule. 2.

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teachers of

CBSETuts.com. All the
solutions given in this
page are solved based
on CBSE Syllabus and
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